¹H AND ¹³C NMR ASSIGNMENTS OF THE THIOPEPTIDE ANTIBIOTIC NOSIHEPTIDE

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The ¹H and ¹³C NMR spectra of nosiheptide have been assigned by use of 2D NMR techniques on unlabeled samples and biosynthetically multiple-labeled samples from stable isotope feeding experiments.

Nosiheptide (1, Fig. 1) is a sulfur-rich antibiotic which was isolated from Streptomyces actuosus^{1,2)}. It is very active in vitro against Gram-positive bacteria³⁾ and used widely as a feed additive for chicken and pigs to increase weight gains⁴⁾. The complete structure of nosiheptide was elucidated by X-ray crystallography¹). First NMR studies on 1 were reported by DEPAIRE et al.^{5,6} in 1977 who partially assigned the ¹³C NMR spectrum.

Extensive use of 2D NMR spectroscopy on both nonlabeled and biosynthetically multiple-labeled samples led to the complete and unambiguous ¹H and ¹³C NMR assignments of nosiheptide, reported in this paper, as a prerequisite for our biosynthetic studies on this compound⁷).





Results and Discussion

¹H NMR Assignments

Several resonance assignments could be made from chemical shift and coupling constant considerations based on first-order inspection of the spectrum. The most upfield signal, a 6.43 Hz doublet at 0.95 ppm, could be assigned to the threonine methyl group and a singlet at 2.63 ppm to the methyl group of the indole moiety. The quartet at 6.46 ppm (J=6.95 Hz) and the doublet at 1.72 ppm (J=6.95 Hz) are assigned to the AX₃ spin system of dehydrobutyrine (But) 3-H and the But methyl group, respectively.

The indole (Ind) moiety shows the expected AMX-type spin system for three adjacent protons with signals at 7.12 ppm (d, J = 7.02 Hz, Ind 5-H), 7.28 (dd, J = 8.32 and 7.01 Hz, Ind 6-H) and 7.60 (d, J = 8.35 Hz, Ind 7-H), although the expected long-range splitting between Ind 5-H and Ind 7-H was not resolved.

The use of 2D homonuclear correlation spectroscopy $(COSY)^{8,9}$ enabled us to assign most of the signals unambiguously, even in congested regions of the spectrum. For example, connectivity patterns determined from COSY shift correlations allowed the sequence of a set of signals starting at 4.09 (dd) and progressing through 1.80 (m) \rightarrow 2.44 (m) \rightarrow 5.63 (dd) \rightarrow 8.35 (d) to be assigned to Glu 4-H, two Glu 3-H, Glu 2-H and Glu NH, respectively. By working through sequential connectivity patterns of different spin systems other groups of signals were assigned to the protons of the cysteine and threonine moieties in a straightforward manner.

Signals for the methylene group of the dehydroalanine (Deala) residue are observed as two singlets at 5.76 and 6.37 ppm, both correlated in the COSY spectrum with the signal at 10.04 ppm (Deala NH). The calculation of chemical shifts according to the additivity rules¹⁰ indicates that the signal of Deala $3-H_E$ is located more downfield than the signal of Deala $3-H_Z$. Attempts to verify these assignments in nuclear Overhauser effect (NOE) experiments by irradiating Deala NH, Deala $3-H_Z$ and Deala $3-H_E$ were inconclusive. The results of the COSY experiment are summarized in Table 1.

Six sharp singlets in the aromatic region represent the protons of the five thiazole (Thz) rings and the pyridine (Pyr) proton. Four of these signals could be assigned by long range (delayed) $COSY^{11}$ correlations. The long range COSY contour plot at 320°K or higher temperatures shows correlations between protons which are 5, 6 and 7 bonds apart, *i.e.*, long range couplings between Thz(5) 5-H (8.55 ppm), Pyr 4-H (7.82) or Thz(4) 5-H (7.88) and Cys 2-H (5.88). The signal at 8.16 ppm which is correlated with the proton of But 3-H at 6.46 ppm was assigned to Thz(2) 5-H. Couplings between protons separated by such large numbers of bonds are favored by the conformational behavior and bond geometry of the molecule, and are not without precedent^{12,13)}. OH and NH protons like Thr OH, Thr NH, Cys NH and Glu NH were identified after exchange with D₂O at 320 and 350°K. Most exchangeable protons showed also COSY correlations. Pyr OH and Deala NH₂ could not be assigned because their signals were not observed in these spectra.

In this way, all proton signals except the two singlets at 8.30 and 8.65 ppm (see below) could be assigned unequivocally (Table 1).

¹³C NMR Assignments

DEPAIRE *et al.*⁵⁾ observed signals for 29 quaternary (sp^2) , 10 methine (sp^2) , 1 methylene (sp^2) , 5 methine (sp^3) , 3 methylene (sp^3) and 3 methyl carbon atoms in the single frequency off resonance decoupled ¹³C NMR spectrum of **1**. An edited distortionless enhancement by polarization transfer (DEPT)¹⁴⁾ analysis

0.95ª	Thr $CH_3 \leftrightarrow 4.00^a$	Thr 3-H	5.76	Deala 3-H _z ↔ 6.37	Deala $3-H_E$
1.72	But $CH_3 \leftrightarrow 6.46$	But 3-H		↔10.04	Deala NH
	↔ 9.32	But NH	5.88	Cys 2-H↔ 3.56	Cys 3-H
1.80	Glu 3-H↔ 4.09	Glu 4-H		↔ 3.86	Cys 3-H
	↔ 5.63	Glu 2-H		↔ 7.72	Cys NH
2.44	Glu 3-H↔ 4.09	Glu 4-H	6.37	Deala 3- $H_E \leftrightarrow 5.76$	Deala $3-H_Z$
	↔ 5.63	Glu 2-H		↔10.04	Deala NH
3.56	Cys 3-H↔ 3.86	Cys 3-H	6.46	But 3-H \leftrightarrow 1.72	But CH ₃
	↔ 5.88	Cys 2-H		↔ 9.32	But NH
3.86	Cys 3-H↔ 3.56	Cys 3-H	7.12	Ind 5-H \leftrightarrow 5.59	Ind 4'-H
	↔ 5.88	Cys 2-H		↔ 7.28	Ind 6-H
4.00	Thr 3-H \leftrightarrow 0.95	Thr CH ₃		↔ 7.60	Ind 7-H
	↔ 4.57	Thr 2-H	7.28	Ind 6-H \leftrightarrow 7.12	Ind 5-H
4.09	Glu 4-H↔ 1.80	Glu 3-H		↔ 7.60	Ind 7-H
	↔ 2.44	Glu 3-H	7.58	Thr $OH \leftrightarrow 4.57$	Thr 2-H
4.57	Thr 2-H \leftrightarrow 4.00	Thr 3-H	7.60	Ind 7-H \leftrightarrow 7.12	Ind 5-H
	↔ 7.58	Thr OH		↔ 7.28	Ind 6-H
	↔ 7.64	Thr NH	7.64	Thr NH \leftrightarrow 4.57	Thr 2-H
5.40	Ind 4'-H \leftrightarrow 5.59	Ind 4'-H	7.72	Cys NH↔ 5.88	Cys 2-H
5.59	Ind 4'-H \leftrightarrow 5.40	Ind 4'-H	8.35	Glu NH↔ 5.63	Glu 2-H
	↔ 7.12	Ind 5-H	9.32	But NH \leftrightarrow 1.72	But CH ₃
5.63	Glu 2-H \leftrightarrow 1.80	Glu 3-H		↔ 6.46	But 3-H
	↔ 2.44	Glu 3-H	10.04	Deala NH↔ 5.76	Deala 3- H_z
	↔ 8.35	Glu NH		↔ 6.37	Deala $3-H_E$

Table 1. 2D NMR COSY shift correlations of nosiheptide.

^a ppm.

Table 2. $-\pi$ C COST correlations of nosineptide at 520 J	Table 2.	¹ H- ¹³ C COS	Y correlations	of nosiheptide	at 320°K.
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Assignment	¹³ C chemical shift (ppm)	¹ H chemical shift (ppm)	Assignment	¹³ C chemical shift (ppm)	¹ H chemical shift (ppm)
Ind CH ₃	12.23	2.63	Thr 3	66.50	4.00
Ind 5	123.23	7.12	Cys 2	49.05	5.88
Ind 6	124.91	7.28	Cys 3	29.49	3.56, 3.86
Ind 7	114.40	7.60	Glu 2	45.15	5.63
Ind 4'	65.90	5.40, 5.59	Glu 3	37.60	1.80, 2.44
Deala 3	103.60	5.76, 6.37	Glu 4	66.40	4.09
But CH ₃	13.50	1.72	Pyr 4	127.12	7.82
But 3	128.85	6.46	Thz(2) 5	124.45	8.16
Thr CH ₃	18.25	0.95	Thz(4) 5	119.98	7.88
Thr 2	56.57	4.57	Thz(5) 5	126.80	8.55

confirmed this result, demonstrating carbon types and numbers to be also consistent with the X-ray analysis¹).

Correlations of the ¹H and ¹³C chemical shifts for 20 of the 22 protonated carbons by their one bond couplings were established through a 2D ¹H-¹³C-COSY¹⁵ experiment as shown in Table 2.

Assignment of most of the quaternary carbons could be achieved by a series of 2D-COLOC¹⁶) experiments optimized for different long-range coupling constants (data summarized in Table 3) and by comparison of the ¹³C NMR data for 1 with literature values for model compounds which had been obtained by acid and alkaline hydrolysis of nosiheptide¹⁷). By comparison with the ¹³C assignments of such a model compound¹⁷), resonance signals of the indole moiety could be assigned to Ind C-3 (118.35 ppm), Ind C-3a (124.70), Ind C-4 (129.20), Ind C-2 (130.40) and Ind C-7a (137.60). The assignments of Ind C-3,

¹³ C assignment (ppm)				
4 Hz	6 Hz	9 Hz ^a		
	Ind C-3 (118.35)	Ind C-3 (118.35)		
Ind C-2 (130.40)	Ind C-2 (130.40)	Ind C-2 (130.40)		
Deala C-2 (134.26)	Deala C-2 (134.26)	Deala C-2 (134.26)		
Deala CO (165.00)	Deala CO (165.00)	Deala CO (165.00)		
Pyr C-4 (127.12)		Pyr C-4 (127,12)		
Pyr C-2 (135.00)	Pyr C-2 (135.00)	Pyr C-2 (135.00)		
	Pyr C-6 (142.52)	Pyr C-6 (142.52)		
	Thz(1) C-2 (163.85)	Thz(1) C-2 (163.85)		
	Ind C-7a (137.60)			
Thz(4) C-4 (153.10)	Thz(4) C-4 (153.10)	Thz(4) C-4 (153.10)		
	Thz(4) C-2 (168.98)	Thz(4) C-2 (168.98)		
Thz(2) C-4 (147.62)	Thz(2) C-4 (147.62)	Thz(2) C-4 (147.62)		
	Thz(2) C-2 (166.32)	Thz(2) C-2 (166.32)		
Thz(3) C-4 (148.70)	Thz(3) C-4 (148.70)	Thz(3) C-4 (148.70)		
	Thz(3) C-2 (170.04)	Thz(3) C-2 (170.04)		
Thz(5) C-4 (149.57)	Thz(5) C-4 (149.57)	Thz(5) C-4 (149.57)		
	Thz(5) C-2 (167.10)	Thz(5) C-2 (167.10)		
Thz(1) C-4 (149.83)	Thz(1) C-2 (163.85)			
	4 Hz Ind C-2 (130.40) Deala C-2 (134.26) Deala CO (165.00) Pyr C-4 (127.12) Pyr C-2 (135.00) Thz(4) C-4 (153.10) Thz(2) C-4 (147.62) Thz(3) C-4 (148.70) Thz(5) C-4 (149.57) Thz(1) C-4 (149.83)	$\begin{array}{c c c c c c c c c c c c c c c c c c c $		

Table 3. COLOC correlations of ¹³C-¹H long range couplings of nosiheptide.

^a The coupling constant values for the 3 COLOC experiments given represent the average long-range C-H coupling constant to which the experiment was tuned and should be interpreted to indicate true values of the coupling constants. The range of the J_{CH} found in the COLOC experiments are consistent with actual values measured in the gated ¹³C NMR spectrum.

Ind C-2 and Ind C-7a are confirmed by two-bond and three-bond coupling correlations. The Deala C-2 and Deala CO resonances can also be differentiated by long range correlations (COLOC): Deala C-2 resonates at 134.26 ppm while Deala CO appears at 165.00.

Among the four quaternary carbon atoms in the pyridine moiety, signals at 129.90, 135.00 and 142.52 ppm can be assigned to Pyr C-5, Pyr C-2 and Pyr C-6, respectively, based on the ¹H-¹³C long-range coupling patterns (3 bond coupling) and the ¹³C NMR data of a model compound¹⁷.

The COLOC correlations within each thiazole ring indicate that the Thz 5-H displays a two-bond coupling with Thz C-4 and a three-bond coupling with Thz C-2. As a consequence, the five signals at 153.10 (Thz(4)), 149.83 (Thz(1)), 149.57 (Thz(5)), 148.70 (Thz(3)) and 147.62 ppm (Thz(2)) are assigned to the carbons of the Thz C-4 groups as well as another five signals at 170.04 (Thz(3)), 168.98 (Thz(4)), 167.10 (Thz(5)), 166.32 (Thz(2)) and 163.85 ppm (Thz(1)) to the Thz C-2 carbons. The ¹³C resonance at 163.85 ppm was assigned to Thz(1) C-2, based on its long-range coupling with Pyr 4-H at 7.82 ppm. An additional long-range correlation of Thz(1) C-2 at 163.85 ppm with the proton signal at δ 8.65 allowed the latter signal to be assigned to Thz(1) 5-H. Therefore, the only remaining signal in the proton spectrum can be assigned to Thz(3) 5-H (8.30 ppm).

The four carbonyl carbons adjacent to the thiazole rings could not be differentiated by the aforementioned methods, and we turned to biosynthetically ¹³C-labeled samples to help assign these additional signals. Our biosynthetic results had shown that the thiazole rings derive from L-cysteine and the carboxyl group of another amino acid⁷. We therefore fed L-[1,2-¹³C₂]serine, a precursor of cysteine. The sample of 1 from this experiment showed the expected enrichment and coupling of the Thz C-4 carbons (from carbon 2 of serine) and the adjacent carbonyl carbons (from carbon 1 of serine). A 2D-INADEQUATE^{18~20}) experiment gave correlations between the previously assigned Thz C-4 carbons

and the carbonyl carbons, which allowed an easy assignment of the latter.

Thz(1) C-4 (149.83 ppm)↔(159.45 ppm) Thz(1) CO

Thz(2) C-4 (147.62 ppm)↔(159.60 ppm) Thz(2) CO

Thz(3) C-4 (148.70 ppm)↔(159.80 ppm) Thz(3) CO

Thz(5) C-4 (149.57 ppm)↔(158.20 ppm) Thz(5) CO

Additional correlations were observed which confirmed assignments already established.

Another feeding experiment with L-[2,3-¹³C₂]serine took advantage of the fact that serine is also incorporated into the pyridine⁷⁾, threonine and glutamic acid moieties[†]. The correlations in the 2D-INADEQUATE spectrum allowed the assignments of Pyr C-3 (from carbon 3 of serine), Thr CO and Glu CO (both from carbon 2 of serine).

Pyr C-2 (135.00 ppm)↔(150.80 ppm) Pyr C-3

Thr CO (167.69 ppm)↔(56.57 ppm) Thr C-2

Glu CO (172.62 ppm)↔(66.40 ppm) Glu C-4

Again, additional correlations were observed which confirmed assignments already made based on other data discussed above.

The two remaining signals in the ¹³C NMR spectrum can be assigned based on their chemical shifts to But C-2 (129.27 ppm) and Ind CO (181.80 ppm).

The complete ¹H and ¹³C NMR assignments for nosiheptide are summarized in Table 4.

Ind CO	181.80 s	But 3	128.85 d	6.46 (q, $J = 6.95$)
Glu CO	172.62 s	Pyr 4	127.12 d	7.82 (s)
Thz(3) 2	170.04 s	Thz(5) 5	126.80 d	8.55 (s)
Thz(4) 2	168.98 s	Thz(1) 5	125.98 d	8.65 (s)
Thr CO	167.69 s	Thz(3) 5	125.25 d	8.30 (s)
Thz(5) 2	167.10 s	Ind 6	124.91 d	7.28 (dd, $J = 7.01, 8.32$)
Thz(2) 2	166.32 s	Ind 3a	124.70 s	
Deala CO	165.00 s	Thz(2) 5	124.45 d	8.16 (s)
Thz(1) 2	163.85 s	Ind 5	123.23 d	7.12 (d, $J = 7.02$)
Thz(3) CO	159.80 s	Thz(4) 5	119.98 d	7.88 (s)
Thz(2) CO	159.60 s	Ind 3	118.35 s	
Thz(1) CO	159.45 s	Ind 7	114.40 d	7.60 (d, $J = 8.35$)
Thz(5) CO	158.20 s	Deala 3	103.60 t	6.37 E (s), 5.76 Z (s)
Thz(4) 4	153.10 s	Thr 3	66.50 d	4.00 (m)
Pyr 3	150.80 s	Glu 4	66.40 d	4.09 (dd, J = 2.24, 11.89)
Thz(1) 4	149.83 s	Ind 4'	65.90 t	5.59 (d, $J = 11.75$), 5.40 (d, $J = 11.71$)
Thz(5) 4	149.57 s	Thz 2	56.57 d	4.57 (dd, $J = 3.75, 7.55$)
Thz(3) 4	148.70 s	Cys 2	49.05 t	5.88 (ddd, $J = 4.89$, 5.21, 9.67)
Thz(2) 4	147.62 s	Glu 2	45.15 d	5.63 (dd, $J = 2.37$, 11.86)
Pyr 6	142.52 s	Glu 3	37.60 t	2.44 S (m), 1.80 R (m)
Ind 7a	137.60 s	Cys 3	29.49 t	3.86 (dd, J = 4.89, 13.86), 3.56 (dd, J = 5.49, 13.95)
Pyr 2	135.00 s	Thr CH ₃	18.25 q	0.95 (d, $J = 6.43$)
Deala 2	134.26 s	But CH ₃	13.50 q	1.72 (d, $J = 6.95$)
Ind 2	130.40 s	Ind CH ₃	12.23 q	2.63 (s)
Pyr 5	129.90 s			
But 2	129.27 s			
Ind 4	129.20 s			

Table 4. ¹H and ¹³C NMR spectral data of nosiheptide (δ in ppm, J in Hz).

Ind NH 11.19 (s, br); Deala NH 10.04 (s); But NH 9.32 (s); Glu NH 8.35 (d, J=8.42); Glu OH 8.00 (s, br); Cys NH 7.72 (d, J=9.67); Thr NH 7.64 (d, J=8.01); Thr OH 7.58 (s, br).

[†] A detailed discussion of these results will be the subject of a future publication.

Experimental

¹H and ¹³C NMR spectra were acquired in DMSO- d_6 at a field strength of 7.1 T on Bruker WM-300 or IBM AF-300 spectrometers. Samples were prepared in 5 mm tubes, and were heated to 320°K in the probe prior to analysis. Spectra were internally referenced to the solvent resonance.

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